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# PLANT-BASED PLASTICIZERS AND FLAME RETARDANTS FOR P-PVC

### Motivation

- increasing scarcity of fossil resources
- substantial public interest on 'green chemistry'
- stringent regulation regarding use of well-established general purpose phthalate plasticizers
- damaging effects of halogenated or heavy metal based flame retardants on health and environment

### Phosphonylation of FAE

Transesterification and phosphonylation of fatty acids of high-oleic sunflower oil, rape seed

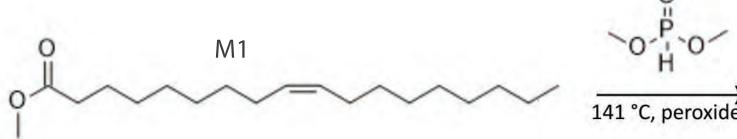
#### Aim

- increasing the thermal stability of P-PVC by use of plant oil based plasticizers
- improvement of compatibility between PVC polymer by increasing the polarity of the fatty acid ester (FAE)
- incorporation of phosphorous functionalities to provide additional thermal stability and flame retardancy

#### Phosphorous plasticizers

The performance potential of phosphonylated FAE (pFAE) was compared to standard





M1MP

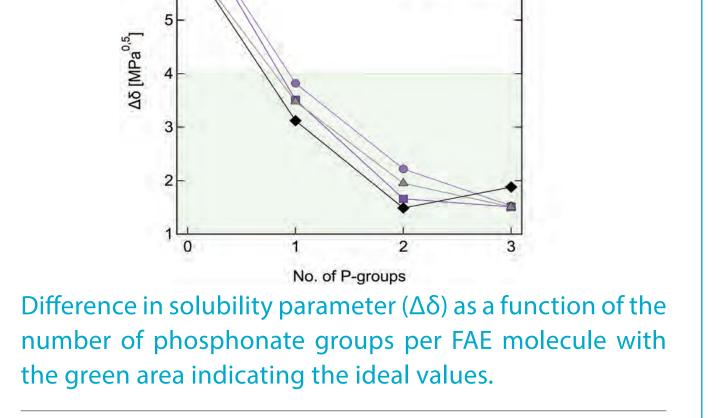
Phosphonylation of high-oleic sunflower oil methyl ester (M1) to a methyl ester of the high-oleic sunflower oil with a methylphosphonate group (M1MP) in place of the double bond.

Solubility parameters ( $\delta$ ) and difference in solubility parameter between plasticizer and PVC ( $\Delta\delta$ )

	δ [MPa <sup>0,5</sup> ]	Δδ [MPa <sup>0,5</sup> ]
PVC	20.3	
DINCH	17.2	5.4
DINP	17.8	4.3
DPO	21.5	2.5
M1MP	19.0	3.1
iP1EP	18.5	3.8

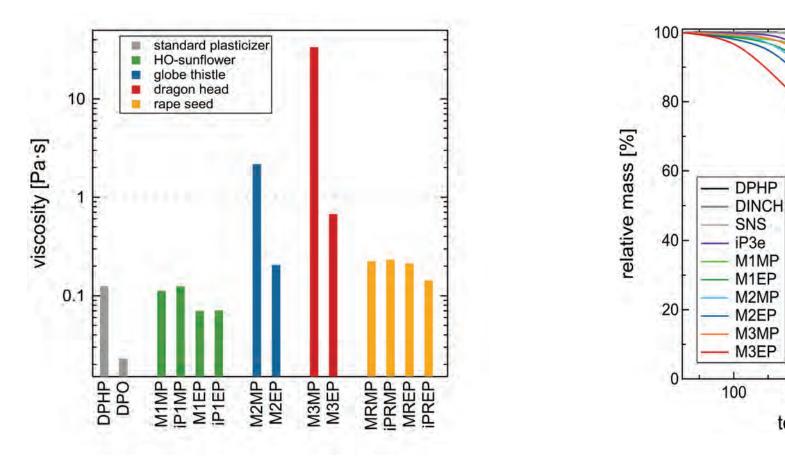
• decrease in difference in solubility

- parameters ( $\Delta\delta$ ) with increasing
- number of phosphonate groups per FAE molecule
- $\Delta\delta$  increases with length of alkyl chains in phosphonate groups as polarity decreases



M-FAE-MP
M-FAE-EP
iP-FAE-MP
iP-FAE-EP

plasticizers and was based on viscosity and thermal behavior (TGA).



Left: Viscosity of the plasticizers depending on their composition. Right: TGA curves showing the reduction in relative mass upon increasing temperature.

- increase in viscosity with increasing No. of P-groups per FAE molecule
- FAE with methylphosphonate groups based on globe thistle and dragons head oil not processible
- multi-step decrease in mass with increasing temperature and > 4 % residual mass for pFAE due to phosphorus content
  - residual mass correlates with theoretical phosphorus content based on the chemical structure

200

300

temperature [°C]

400

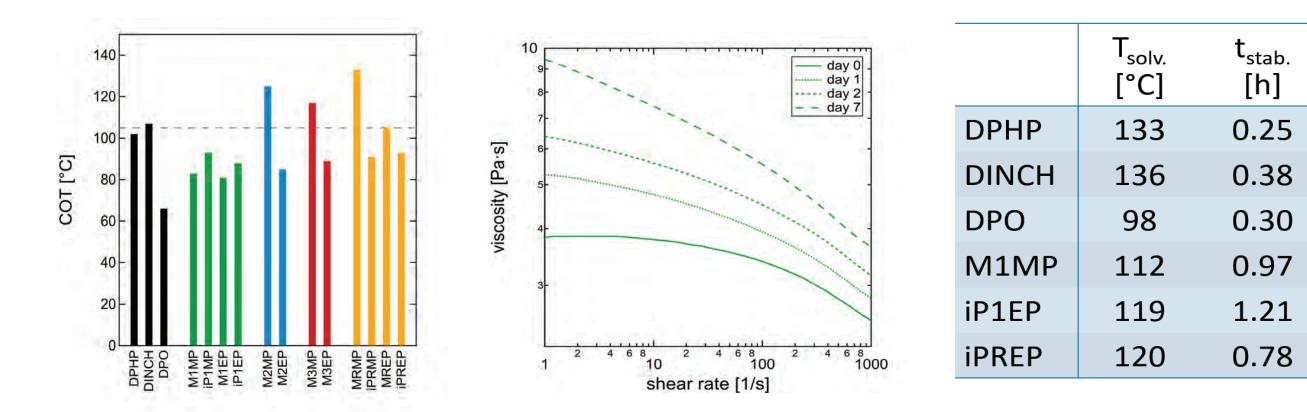


## PVC plastisols with pFAE



## PVC artificial leather with pFAE

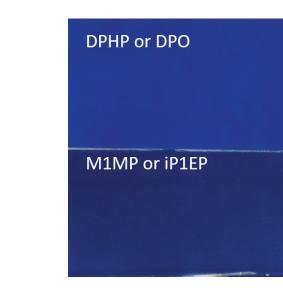
Plastisols with PVC (K value 70), 65 phr plasticizer and 2 phr thermostabilizer were used for the screening for gelation, solution temperature, time-dependent viscosity and thermal stability.

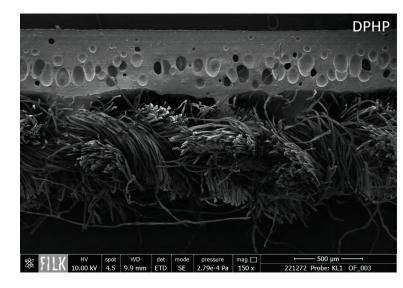


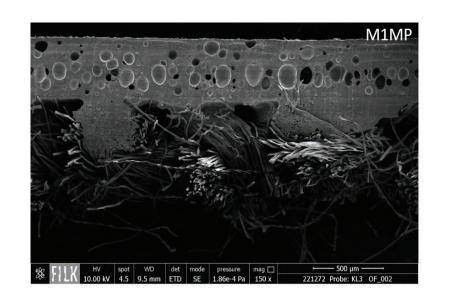
Left: Cross-over-temperatures (COT) from rheological gelation curves. Middle: Viscosity of iP1EP depending on the day on or after plastisol preparation. Right: Solution temperature  $(T_{colv})$  and stability time  $(t_{ctab})$  by dehydrochlorination (similar to ISO 182-3) for selected plasticizers.

- excellent gelation behavior with cross-over-temperatures (COT) of 81-105 °C for most pFAE
- increasing viscosity of plastisol as time after preparation increased
- plastisols with HO-sunflower and rape seed oil based pFAE still processible after 2 days of ripening (viscosity < 10 Pa⋅s)
- solution temperature also demonstrated very good gelation behavior
- delaying dehydrochlorination of plastisols with pFAE

Compact PVC artificial leather consisting of top, intermediate and base layer with pFAE as plasticizer were prepared in a lab coater. Artificial leather with DPHP contained a flame retardant mix with boron and antimony.







Left: Photographs of artificial leather with blue pigment. Middle & Right: SEM images of cross sections through artificial leather with DPHP or M1MP.

- pFAE influence color and foaming of artificial leather
- self-extinguishing behavior of all samples with smaller burning rate for M1MP
- higher thermal stability for artificial leather with M1MP

Results of testing limiting oxygen index (LOI), thermal stability by dehydrochlorination (similar to ISO 182-3), horizontal burning behavior (similar to DIN 75200) and maximum heat release rates by cone-calorimetry; \*- determined before reaching first measuring mark

	DPHP	DPO	M1MP
LOI [% O <sub>2</sub> ]	23.8	25.5	25.8
therm. stability [h]	0.5	0.7	1.2
burning behavior	SE/0	SE/0	SE/0
burning rate* [mm/min]	21	23	19

	peak heat release rate [kW/m <sup>2</sup> ]	203.3

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#### Conclusion

- $\rightarrow$  pFAE are good candidates for application as alternative plasticizers.
- $\rightarrow$  pFAE based on locally sourced plant oils displayed best plasticizer performance and good flame retardancy effects.
- → Favorable lower gelation and solution temperatures of pFAE compared to phthalate  $\rightarrow$  plasticizers.
- → Gelation behavior improved and viscosity decreased with increasing length of alkyl chain in phosphonate group.
- → Smaller differences in solubility parameter between pFAE and PVC was not enough to increase plasticizer performance as viscosity increased significantly.
- $\rightarrow$  Flame retardancy improved by pFAE to such an extend that no additional flame retardants were necessary.

#### Contact

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